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## A [2 + 2] Cycloaddition Dimer and a Diels—Alder Adduct from *Alpinia katsumadai*

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Plants of the genus *Alpinia* (Zingiberaceae) are herbs that have been traditionally used in China and some southeast Asian countries for relieving stomachache, treating colds, invigorating the circulatory system, and reducing swelling.<sup>1</sup> Previous investigations on the genus *Alpinia* led to the isolation of some new diarylheptanoids bearing a chalcone or a flavanone moiety<sup>2-4</sup> and monocyclic sesquiterpenes adducted by a chalcone.<sup>5</sup> In our continuing endeavor to discover new natural products from *Alpinia katsumadai*, a novel katsumadain dimer named katsumadain C (1), and a unique chalcone diarylheptanoid adduct named calyxin Y (2) were isolated from the seeds of this species. Herein, details of the isolation, structural elucidation, and postulated biogenetic origin are described.



Katsumadain C (1)<sup>6</sup> had the molecular formula of  $C_{46}H_{52}O_6$  based on the HRESIMS (m/z 699.3641 [M – H]<sup>-</sup>).

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<sup>(6)</sup> Katsumadain C (1): white amorphous powder;  $[\alpha]^{28}_{D} + 173.2$  (c 0.050, CHCl<sub>3</sub>/MeOH = 1:1); UV (CHCl<sub>3</sub>/MeOH = 1:1) $\lambda_{max}$  (log  $\varepsilon$ ) 237 (2.19), 301 (4.43) nm; IR (KBr)  $\nu_{max}$  3446, 2957, 2934, 1655, 1577, 1409, 1292, 1111, 1022, 839, 752, 699 cm<sup>-1</sup>; <sup>1</sup>H NMR and <sup>13</sup>C NMR data, see Table 1; negative ESI-MS *m*/*z* 699.5 [M - H]<sup>-</sup>; HRESIMS *m*/*z* 699.3641 [M - H]<sup>-</sup> (calcd for C<sub>46</sub>H<sub>51</sub>O<sub>6</sub>, 699.3691).

**Table 1.** <sup>1</sup>H (500 MHz) and <sup>13</sup>C (125 MHz) NMR Data of 1 (in DMSO- $d_6$ )

	unit I		unit II		
no.	$\delta_{\rm H}({\rm mult},J,{\rm Hz})$	$\delta_{\mathrm{C}}$	$\delta_{\rm H}({\rm mult},J,{\rm Hz})$	$\delta_{ m C}$	
1		162.9		162.8	
2		103.7		103.6	
3		164.6		164.6	
4	5.92(s)	100.8	5.88(s)	100.5	
5		160.5		160.4	
6	4.20 (dd, 9.8, 6.3)	44.2	4.26 (dd, 9.8, 6.3)	43.8	
7	4.30 (dd, 9.8, 6.3)	42.8	4.32 (dd, 9.8, 6.3)	42.6	
8		138.0		137.9	
9, 13	7.32 (m)	127.5	7.30 (m)	127.3	
10, 12	7.24 (m)	127.9	7.23 (m)	127.9	
11	7.16 (m)	126.4	7.15 (m)	126.4	
1'		131.9		131.8	
2'	4.85(s)	125.0	4.85(s)	125.0	
3'	$3.33 (m)^a$	35.0	$3.33 (m)^a$	34.9	
4'	$1.84 ({ m m})^a$	39.7	$1.84 \ (m)^a$	39.6	
5'	$1.13  (m)^a$	22.5	$1.12 \ (m)^a$	22.5	
	$1.61  (m)^a$		$1.60 \ (m)^a$		
6'	$1.93  (m)^{a}$	30.3	$1.93 (m)^{a}$	30.3	
	$1.86 ({ m m})^a$		$1.84 (m)^{a}$		
7'	1.53(s)	23.2	1.53(s)	23.1	
8'	$1.19 ({ m m})^a$	27.8	$1.19 (m)^{a}$	27.8	
9'	0.60 (d, 6.8)	16.4	0.60 (d, 6.8)	16.4	
10′	0.75(dd,6.8,1.2)	21.2	$0.75 (\mathrm{dd}, 6.8, 1.2)$	21.2	
3-OH	10.78 (s)		10.78(s)		

<sup>a</sup> Signal pattern unclear due to overlapping.

The IR spectrum indicated the presence of OH ( $3446 \text{ cm}^{-1}$ ), ester carbonyl (1655 cm<sup>-1</sup>) groups, and aromatic rings (1577, 1514, and 1409 cm<sup>-1</sup>). The <sup>1</sup>H, <sup>13</sup>C, and HSQC NMR spectra (Table 1) indicated the presence of a monosubstituted benzene ring ( $\delta_{\rm H}$  7.15–7.33, 5H, overlapped), two methylenes, seven methines (including two olefinic methines), and three angular methyls [ $\delta_{\rm H}$  0.60 (d, J = 6.8 Hz, 3H), 0.75 (dd, J = 6.8, 1.2 Hz, 3H), 1.53 (s, 3H);  $\delta_{\rm C}$  16.4, 21.2, 23.1]. These spectral features were like those of katsumadain<sup>7</sup> except for the absence of trans-olefinic signals, and instead, the presence of two additional methines, which combined with 2-fold relationship of the molecular weight of 1 ( $C_{46}H_{52}O_6$ ) and katsumadain  $(C_{23}H_{26}O_3)$ , suggested 1 is a dimer of katsumadain, forming a cyclobutane ring between the two units. Two possible cycloadducts, head-to-tail or head-to-head, exists (Figure 1), which cannot be distinguished directly by NMR data. On the basis of the mass spectrometric studies on similar situations,<sup>8,9</sup> three typical fragments at m/z 350, 180, and 520 should be found for 1 in head-to-head mode, while only one fragment ion at m/z 350 could be found in head-to-tail mode. The EI-MS of 1 exhibited a very small molecular ion at 700 and the presence of a typical fragment at m/z 350 indicated 1 to be a head-to-tail dimer. In addition, as shown in Figure 1,



**Figure 1.** Two possible dimer modes for **1**. The correlations between the calculated and experimental chemical shifts for the head-to-head mode and the head-to-tail mode.

the closer fitting between the predicted and experimental <sup>13</sup>C NMR chemical shifts for the head-to-tail mode (R = 0.997) than the head-to-head mode (R = 0.988) also indicate the head-to-tail mode is the preferred one.<sup>10</sup>



Figure 2. Selected HMBC  $(\rightarrow)$  and ROESY  $(\leftrightarrow)$  correlations of 1.

In the HMBC spectrum (Figure 2), the proton signal of H-6 ( $\delta_{\rm H}$  4.20) was correlated with the carbon signals of C-7 ( $\delta_{\rm C}$  42.8), and the proton signals of H-7 ( $\delta_{\rm H}$  4.30) with the carbon signals at C-6 ( $\delta_{\rm C}$  44.2). The HMBC spectrum showed long-range correlations from the protons of the monosubstituted benzene ring at  $\delta_{\rm H}$  7.30 (H-9 and H-13) to the carbon signals of C-7. Thus, the position of the monosubstituted benzene ring was determined to be at C-7 position. The proton signal at  $\delta_{\rm H}$  4.20 (H-6) was correlated with the carbon signals in lactonic ring ( $\gamma$ -pyrone) at  $\delta_{\rm C}$  100.8 (C-4) and 160.5 (C-5). Therefore, the planar structure of **1** was established as shown in Figure 1. The relative configuration

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of compound 1 was deduced from the analysis of its ROESY correlations (Figure 2) and the energy minimized molecular modeling using density functional theory (DFT) at the B3LYP/6-31G+(d,p) basis set level in Gaussian 09.<sup>11</sup> The ROESY correlation from H-3' to H-8' and OCH<sub>3</sub>-9' confirmed the partial stereostructure of 1, which was consistent with katsumadain.<sup>7</sup> The key points were the assignments of the relative coupling system consisting of at least sixteen symmetrical peaks<sup>8,12</sup> was found for the four methine protons on the cyclobutane ring in the <sup>1</sup>H NMR spectrum, suggested a configurations of the cyclobutane ring. A typical AA'BB' symmetrically substituted cyclobutane ring. A 2D J-resolved experiment determined the  ${}^{3}J$  coupling constants of H–I-6/ H-I-7 and H-II-6/H-I-7 to be 6.3 and 9.8 Hz, respectively, which implied that 1 to be a trans-trans fused dimer in headto-tail mode. The computer modeled 3D structure analysis of 1 was compatible with the aforementioned relative configuration as shown in Figure 2, and thus established the whole stucture except for the absolute configuration.

Table 2. <sup>1</sup>H (600 MHz) and <sup>13</sup>C (150 MHz) NMR Data of 2 (in DMSO- $d_6$ )

no.	$\delta_{\rm H}({\rm multi},J,{\rm Hz})$	$\delta_{\mathrm{C}}$	no.	$\delta_{\rm H}({\rm multi},J,{\rm Hz})$	$\delta_{\mathrm{C}}$
1	$2.84 (\mathrm{m,2H})^a$	30.2	2'''		163.1
2α	$3.05 (m)^a$	38.5	3'''	5.81 (d, 1.8)	91.7
$2\beta$	$3.28  (m)^a$				
3		199.9	4'''		165.6
4		137.6	5'''	5.91 (d, 1.8)	96.4
5	7.31 (br s)	147.1	6'''		167.6
6	$4.41({\rm br}~{\rm dd},10.2,7.2)$	64.8	7'''		203.5
7	3.11 (dd, 10.2, 4.2)	47.1	8'''	4.75(s)	51.5
1'		141.7	9'''	3.18 (d, 4.2)	49.5
2', 6'	7.28 (m)	128.8	10'''		141.3
3', 5'	7.28 (m)	128.7	$11^{\prime\prime\prime},15^{\prime\prime\prime}$	6.47 (d, 7.2)	128.7
4'	7.18 (m)	126.4	$12^{\prime\prime\prime}, 14^{\prime\prime\prime}$	7.09 (m)	127.9
$1^{\prime\prime}$		140.5	13'''	7.15(m)	126.9
$2^{\prime\prime}, 6^{\prime\prime}$	6.67 (m)	129.2	$OCH_3$	2.89(s)	55.6
$3^{\prime\prime},5^{\prime\prime}$	7.10 (m)	128.1	6-OH	5.24 (d, 7.2)	
$4^{\prime\prime}$	7.12(m)	126.6	$4^{\prime\prime\prime}-OH$	10.70(s)	
$1^{\prime\prime\prime}$		103.5	$6^{\prime\prime\prime}-{\rm OH}$	13.63(s)	

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(13) Calyxin Y (2): white amorphous powder;  $[\alpha]^{28}_{\rm D}$  +2.75 (c 0.080, CHCl<sub>3</sub>/MeOH = 1:1); UV (CHCl<sub>3</sub>/MeOH = 1:1)  $\lambda_{\rm max}$  (log  $\varepsilon$ ) 238 (3.17), 292 (3.20) nm; IR (KBr)  $\nu_{\rm max}$  3445, 2922, 2851, 1631, 1463, 1379, 1108, 810, 720, 696 cm<sup>-1</sup>; <sup>1</sup>H NMR and <sup>13</sup>C NMR data, see Table 2; negative ESIMS m/z 547.3 [M - H]<sup>-</sup>; HRESIMS m/z 547.2108 [M - H]<sup>-</sup> (calcd for C<sub>35</sub>H<sub>31</sub>O<sub>6</sub>, 547.2126).

Calyxin Y  $(2)^{13}$  was isolated as white amorphous powder. Its molecular formula of  $C_{35}H_{32}O_6$  was determined by the observed ion at m/z 547.2108 [M – H]<sup>-</sup> in HRESIMS, which indicated 20 degrees of unsaturation. The maximum UV absorptions at 237 and 292 nm indicated the presence of a conjugated system. In the IR spectrum, absorption bands at  $3700-3200 \text{ cm}^{-1}$  (hydroxyl group) and 1631, 1463, 720, 696 cm<sup>-1</sup> (aromatic ring) were observed. The <sup>13</sup>C NMR spectra (Table 2) resolved 35 carbon resonances that came from three monosubstituted phenyl groups, a methoxyl group, two methylenes, four sp<sup>3</sup> methines, two aromatic protons, an olefinic methine, and two ketone carbonyls. The benzene ring with two aromatic protons [ $\delta_{\rm H}$  5.81 (1H, d, J = 1.8 Hz), 5.91 (1H, d, J = 1.8 Hz);  $\delta_{\rm C}$  91.7, 96.4], the ketone carbonyls  $[\delta_{\rm C} 203.5]$ , two sp<sup>3</sup> methines  $[\delta_{\rm H} 3.18, 4.75; \delta_{\rm C} 49.5, 51.5]$ , and a monosubstituted phenyl group formed a dihydrochalcone moiety. Its structure was indicated by the HMBC correlations (Figure 3) from the methine protons at  $\delta_{\rm H}$ 3.18 (H-9''') to the aromatic carbons at  $\delta_{\rm C}$  141.3 (C-10'''), 128.7 (C-11<sup>'''</sup>, C-15<sup>'''</sup>), and to the ketone carbon at  $\delta_{\rm C}$  203.5 (C-7"). The position of OMe at C-2" was confirmed by the correlation between the proton signals of methoxy group and C-1"", C-2", C-3". The correlation between H-8" and H-9<sup>'''</sup> was observed in the  ${}^{1}H-{}^{1}H$  COSY spectrum. The remaining 19 signals including two other phenyl rings formed a diarylheptanoid moiety. The long-range HMBC correlations H-2'/C-1, H-1/C-3, H-5/C-3, H-6/C-7, H-2"/C-7, and the <sup>1</sup>H-<sup>1</sup>H COSY correlation H-1/H-2, H-5/H-6, H-6/H-7 allowed us to assign the diarylheptanoid part. A hydroxyl signal at  $\delta_{\rm H}$  5.24 (d, J = 7.2 Hz) has a correlation with the signals at  $\delta_{\rm C}$  47.0 (C-7), 64.8 (C-6) and 147.1 (C-5), which means the hydroxyl group located at C-6. These data indicated 2 to be an adduct of a chalcone and a diarylheptanoid.



**Figure 3.** Key HMBC  $(\rightarrow)$ , <sup>1</sup>H-<sup>1</sup>H COSY( $\frown$ ), and ROESY  $(\leftrightarrow)$  correlations for **2**.

The above-mentioned groups and structural fragments, four benzene rings, a double bond and two ketone carbonyls represented 19 degrees of unsaturation. The remaining one degree of unsaturation indicated compound **2** to have another ring. The <sup>1</sup>H $^{-1}$ H COSY correlations (H-6/H-5 and H-7; H-9<sup>'''</sup>/H-7 and H-8<sup>'''</sup>) and HMBC correlations (H-5/C-7 and C-8<sup>'''</sup>; H-6/C-7; H-7/C-8<sup>'''</sup>; OH-6/C-5 and C-7) established a cyclohexene ring with a hydroxyl at C-6 between diarylheptanoid and chalcone. Thus, the planar structure of **2** was established as shown in Figure 3.

The relative configuration of 2 was elucidated by NOESY experiment (Figure 3). The intense correlations between H-8"' ( $\delta_{\rm H}$  4.75) and H-9"' ( $\delta_{\rm H}$  3.18) showed that they were cofacial and were arbitrarily assigned as  $\alpha$ -oriented. The mutual correlations from H-9<sup>'''</sup>, H-6, and H-7 to H-2"/6" and from H-8", H-9" and H-6 to H-11"// 15" indicated that the two benzene rings at C-9" and C-7, and H-6 were on the same side, in  $\beta$ -orientation (Figure 3). Noteworthily, signals for OMe ( $\delta_{\rm H}$  2.89) and the two phenyl groups ( $\delta_{\rm H}$  6.67, 6.47) at C-7 and C-9" were severely upfield shifted due to the strong shielding effect of magnetic anisotropy, which was consistent with the observed NOE correlations of OMe/H-11" (or H-15") and H-2''/H-13'''. Thus, the structure of calyxin Y (2) except for the absolute configuration was established as shown in Figure 3.

Katsumadain C (1) represented the first monoterpene substituted kavalactone dimer, conjugated in a head-to-tail mode; calyxin Y (2) was a unique chalcone-diarylheptanoid Diels-Alder adduct with a novel carbon framework of a

Scheme 1. Plausible Biogenetic Pathway of 1



cyclohexene ring, rather than of a tetrahydropyran ring such as blepharocalyxins A and B from *Alpinia blepharocalyx*.<sup>14,15</sup>

Since the two new adducts and key intermediates, diarylheptanoid,<sup>16</sup> chalcone,<sup>17</sup> and katsumadain,<sup>7</sup> were all found in the same plant, we could tentatively outline plau-

sible biogenetic relationships of the isolates. Katsumadain C (1) may be derived through a [2 + 2] cycloaddition reaction of two ethylenic bonds between two katsumadain molecules (Scheme 1). Calyxin Y (2) could be formed through a Diels–Alder addition reaction between a diarylheptanoid and a chalcone (Scheme 2).





The cytotoxicities of katsumadain C (1) and calyxin Y (2) were evaluated against human tumor cell lines A375 (human melanoma cell line), MCF-7 (human breast cancer cell line), SMMC-7721 (human hepatic liver carcinoma cell line), and HCT-116 (human colon carcinoma cell line), using 5-fluorouracil as a positive control with IC<sub>50</sub> values at 20.4, 33.8, 26.8, and 50.0  $\mu$ M, respectively. They generally showed marginal to moderate activities. Compounds 1 and 2 both exhibited significant growth inhibitory effects against SMMC-7721 cells, with IC<sub>50</sub> at 4.8, 9.7  $\mu$ M, respectively, comparable to 5-fluorouracil.

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**Supporting Information Available.** Experimental procedures; IR, ESIMS, HRESIMS, and 1D and 2D NMR spectra of katsumadain C (1) and calyxin Y (2). This material is available free of charge via the Internet at http://pubs.acs.org.

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